

Cis- and Trans-Arrangements of Bis- μ -Acetato Groups in Molybdenum(II) Dimers: Crystal Structures of $\text{Mo}_2(\text{O}_2\text{CMe})_2(\text{MeCOCHCOMe})_2$ and $(\text{Ph}_4\text{As})_2[\text{Mo}_2(\text{O}_2\text{CMe})_2\text{Cl}_4] \cdot 2\text{MeOH}$

C. DAVID GARNER*, STEPHEN PARKES and IAN B. WALTON

Department of Chemistry, Manchester University, Manchester, M13 9PL, U.K.

WILLIAM CLEGG**†

Department of Chemistry, Newcastle University, Newcastle-upon-Tyne, NE1 7RU, U.K.

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The potential versatility in the co-ordination chemistry for a pair of metal atoms held together by a strong metal–metal bond is gradually being realised in the progression of synthetic and structural studies reported for such systems [1, 2]. In particular, complexes of Mo_2^{4+} are currently receiving much attention. $\text{Mo}_2(\text{O}_2\text{CMe})_4$ was the first such compound to be prepared [3] and characterised [4] and studies on this and related tetra- μ -carboxylato-complexes have been important in the understanding of the strength [5] and structure [6] of metal–metal quadruple bonds. In order to pursue certain of these studies in more detail we have prepared a number of complexes containing carboxylate and other ligands co-ordinated to Mo_2^{4+} . Here we report the isolation and X-ray crystallographic characterisation of $(\text{Ph}_4\text{As})_2[\text{Mo}_2(\text{O}_2\text{CMe})_2\text{Cl}_4] \cdot 2\text{MeOH}$ (I) and $\text{Mo}_2(\text{O}_2\text{CMe})_2(\text{pd})_2$ (II) (pd = 2,4-pentanedionate).

Preparation

All manipulations were carried out under an atmosphere of purified dinitrogen. I was prepared by reacting $\text{Mo}_2(\text{O}_2\text{CMe})_4$ with an excess of Ph_4AsCl in dilute hydrochloric acid. Crystals suitable for X-ray crystallographic studies were obtained by recrystallisation from Analar MeOH. II was prepared by the reaction of $\text{Mo}_2(\text{O}_2\text{CMe})_4$ with $\text{Na}(\text{pd})$ in thf under reflux. The subsequent concentration and cooling of this mixture resulted in the precipitation of a pale-brown powder. Sublimation of this material afforded yellow crystals of $\text{Mo}_2(\text{O}_2\text{CMe})_4$ and a small quantity of brown crystals which were separated and

*Authors for correspondence.

†Present address: Anorganisch-Chemisches Institut der Universität Göttingen, D3400 Göttingen, F.R.G.

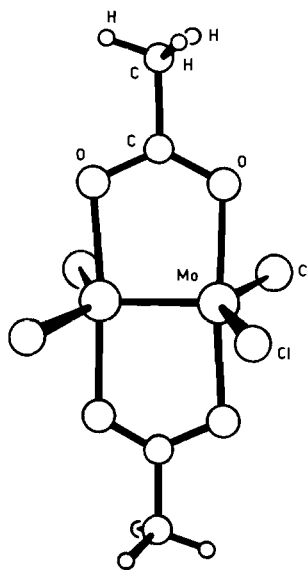


Figure 1. Atomic arrangement of the anion in $(\text{Ph}_4\text{As})_2[\text{Mo}_2(\text{OAc})_2\text{Cl}_4] \cdot 2\text{MeOH}$.

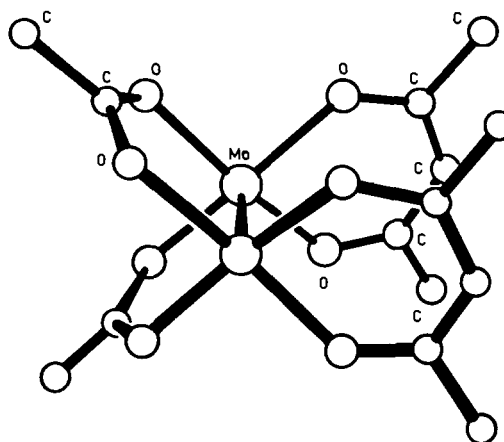


Figure 2. Molecular structure of $\text{Mo}_2(\text{OAc})_2(\text{pd})_2$ (hydrogen atoms omitted for clarity).

used for X-ray crystallographic measurements. II was also prepared by the reaction of I with $\text{Na}(\text{pd})$ in EtOH.

Crystal Data

$\text{C}_{54}\text{H}_{54}\text{O}_6\text{As}_2\text{Mo}_2$ (I), triclinic, $a = 10.571$, $b = 11.330$, $c = 13.397$ Å; $\alpha = 75.16$, $\beta = 67.19$, $\gamma = 64.24^\circ$. Space group $P1$, $Z = 1$, $R = 0.0575$ (0.0726 weighted) for 2165 independent reflections.

$\text{C}_{15}\text{H}_{20}\text{O}_8\text{Mo}_2$ (II), monoclinic, $a = 8.041$, $b = 12.373$, $c = 18.851$; $\beta = 102.02^\circ$. Space group $P2_1/c$, $Z = 4$, $R = 0.0335$ (0.0434 weighted) for 1806 independent reflections.

The atomic arrangements about the Mo_2^{4+} centre in the compounds I and II are illustrated in Figures 1 and 2, respectively. The $[\text{Mo}_2(\text{O}_2\text{CMe})_2\text{Cl}_4]^{2-}$ anion has the *trans* configuration required since the moiety is located on a crystallographic centre of symmetry. The framework of this complex closely corresponds to D_{2h} point symmetry with dimensions, $\text{Mo}-\text{Mo} = 2.086(2)$, $\text{Mo}-\text{O} = 2.120(12)$, $\text{Mo}-\text{Cl} = 2.434(4)$ Å, and $\text{Cl}-\text{Mo}-\text{Cl} = 141.8(1)^\circ$. The methanol molecules of the compound I are not co-ordinated to the molybdenum atoms. The constituent molecules of II have a *cis* configuration of the two μ -acetato-groups and two bidentate pentanedionato-groups. The molecular framework closely approximates to C_{2v} point symmetry with dimensions $\text{Mo}-\text{Mo} = 2.129(1)$, $\text{Mo}-\text{O}(\text{OAc}) = 2.10(2)$, $\text{Mo}-\text{O}(\text{pd}) = 2.09(1)$ Å, and $\text{O}(\text{pd})-\text{Mo}-\text{O}(\text{pd}) = 86.0(4)^\circ$. There are $\text{O}(\text{OAc})\cdots\text{Mo}$ intermolecular contacts, approximately co-linear with the metal-metal axis, of length 2.709 Å. The dimensions of the bridging acetato-groups of I and II and the pentanedionato-groups of II are typical of the corresponding dimensions reported for other structure determinations [1, 2, 7].

Discussion

The intermetallic distances in these complexes are typical of those observed for other $\text{Mo}^{\text{II}}-\text{Mo}^{\text{II}}$ quadruple bonds [1]; the slightly longer bond in II *versus* I probably being associated [2] with the $\text{Mo}\cdots\text{O}(\text{OAc})$ intermolecular contacts of II whereas there is no comparable axial binding for I. Although *cis* and *trans* arrangements of a pair of μ -acetato-groups have been characterised for a number of complexes with a dimetal centre, the ready synthesis of II from I illustrates the flexible co-ordination

chemistry of such centres. The characterisation of II is of particular interest since it is the first time that a 2,4-pentanedionatogroup has been characterised in studies of Mo_2^{4+} and related complexes.

Acknowledgment

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